This article was downloaded by:

On: 29 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

SYNTHESIS AND REACTIONS OF SOME ISOQUINOLINE DERIVATIVES

A. M. El-khawaga^a; G. M. El-naggar^a; Kh. M. Hassan^a; A. M. Kamal El-Dean^a
^a Department of Chemistry, Faculty of Science, Assiut University, Assiut, Egypt

To cite this Article El-khawaga, A. M. , El-naggar, G. M. , Hassan, Kh. M. and El-Dean, A. M. Kamal(1989) 'SYNTHESIS AND REACTIONS OF SOME ISOQUINOLINE DERIVATIVES', Phosphorus, Sulfur, and Silicon and the Related Elements, 44:3,203-207

To link to this Article: DOI: 10.1080/10426508908040610 URL: http://dx.doi.org/10.1080/10426508908040610

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

SYNTHESIS AND REACTIONS OF SOME ISOQUINOLINE DERIVATIVES

A. M. EL-KHAWAGA*, G. M. EL-NAGGAR, Kh. M. HASSAN, and A. M. KAMAL EL-DEAN

Department of Chemistry, Faculty of Science, Assiut University, Assiut, Egypt.

(Received August 25, 1988; in final form November 3, 1988)

Treatment of 3,4-tetramethylene-6-amino-5-cyano-thiopyran-2(1H)thione (1) with alkali followed by acidification with HCl afforded 1,2,5,6,7,8-hexahydro-4-cyano-3(2H)-oxoisoquinoline-1-thiol (2). When compound (1) was reacted with excess alkyl halides the corresponding O- and S-alkylated isoquinolines (3) were obtained. Also, α -chloroacetic acid and β -bromopropionic acid reacted with compound (2) to produce thiazolo and thiazinoisoquinolines (4) and (5). Furthermore, the reaction of (2) with one mole of phenacylbromide produced-S-phenacyl derivatives (6), which cyclized to give the corresponding thiazoloisoquinoline (7). However two moles of phenacyl bromides gave the corresponding S- and O-phenacyl derivatives (8). The later compounds were cyclized to give furoisoquinoline derivative (9).

Key words: Thiopyranthione; cyclohexylidene malonitrile-, isoquinoline derivatives.

Isoquinoline derivatives were reported¹⁻⁴ to act as inhibitors of the enzyme phenylethanolamine-N-methyltransferase (PNMT). PNMT catalyzes the final step in the biosynthesis of epimephrine, effecting the transfer of methyl group from S-adenosylmethionine to the terminal nitrogen of norepinephrine. These compounds are of biological interest as potential therapeutic agents for disorders where specific control of adrenal epinephrine production might be beneficial.⁵

Considering the foregoing benefits we aimed to synthesize isoquinoline thiol as a starting material, then subjecting it to the reaction of different reagents to get some new heterocyclic compounds containing the isoquinoline moiety.

RESULTS AND DISCUSSION

Treatment of cyclohexanone with malononitrile and CS_2 afforded 3,4-tetramethylene-6-amino-5-cyano-2(1H)-thiopyranethione(1). When (1) was subjected to the effect of alkali followed by acidification, the corresponding tetrahydro isoquinoline thiol (2) was obtained. When compound (2) was treated

with methyl and/or ethyl iodide, alkylation occurred at both oxygen and sulphur and the dialkyl derivatives (3a-b) were obtained

Beside its easy alkylation, compound (2) was considered as a versatile intermediate for building up a third angular ring to produce some interesting tricyclic compounds. Thus, when (2) was treated with chloroacetic and/or β -bromopropionic acid in refluxing ethanol and sodium acetate, 2,3,7,8,9,10-hexahydro-3,5-dioxo-5H-thiazolo[2,3-a]-isoquinoline-6-carbonitrile(4), and 3,4,8,9,10,11-hexahydro-4,6-dioxo-2H,6H[1,3]thiazino[2,3-a]isoquinoline-7-carbonitrile(5) were obtained respectively. The reaction pathway is expected to proceed as follows:

The formation of compounds (4) and (5) indicates that compound (2) is preferentially S-alkylated than O-alkylated if there is enough alkylating agent. Evidence in support of this behavior comes from the reaction of compound (2) with phenacyl bromide derivatives. Thus when compound (2) was treated with one mole of phenacyl bromide derivatives, the S-alkylation product, 2,3,5,6,7,8-hexahydro-1-aroylmethylthio-3-oxoisoquinoline-4-carbonitrile(6a-c) were obtained.

The produced compound (6a) was readily dehydrated upon treatment with acetic anhydride or heating over its melting point to give the corresponding

thiazoloisoquinoline derivatives (7) as follows:

$$\begin{array}{c}
CN \\
O \\
SCH_2 - \overline{C} - Ar
\end{array}$$

$$\begin{array}{c}
C_2O \\
S \\
S \\
\end{array}$$

$$\begin{array}{c}
CN \\
O \\
S \\
\end{array}$$

However when compound (2) was reacted with two moles of phenacyl bromide, both S- and O-alkylation occurred and the corresponding dialkylation products (8a-c) were obtained.

O-alkylation of (2) was confirmed by refluxing an ethanol solution of compound (8a) in presence of K_2CO_3 where cycloaddition on the cyano group has occurred to produce the corresponding furoisoquinoline (9).

EXPERIMENTAL

All chemicals used were reagent grade and were purified prior to use. Melting points were determined on kofler melting point apparatus and are uncorrected. Elemental analysis was performed on Perkin-Elmer 240E Microanalyzer. IR spectra were recorded on a Pye-Unicam infrared spectrophotometer using KBr wafer technique. NMR spectra were recorded on NT-200 spectrometer in the suitable deuterated solvent using TMS as internal standard. Mass spectra were determined on Dupont 21-492 B mass spectrometer at an ionizing potential 75 ev, ionizing current $300\,\mu\text{A}$, and source temperature 200°C .

3,4-Tetramethylene-6-amino-5-cyano-2(1H)thiopyranethione(1). To a mixture of 6.6 gm (0.1 mole)-malononitrile, 10 ml cyclohexanone (0.1 mole), 20 ml carbondisulphide, 30 ml methanol, and 5 ml dimethylformamide, 6 ml of triethylamine was added dropwise. The mixture was stirred at room temperature until the product starts to precipitate. The reaction mixture was allowed to stand for 24 hours. The solid product was then filtered off, washed well with alcohol and recrystallized from propanol, m.p. 270-2°C, Lit 272°C.6

Anal. Cacd: $C_{10}H_{10}N_2S_2$: C, 54.05; H, 4.50; N; 12.61; S, 28.82%. Found: C, 54.45; H, 4.22; N, 12.32; S, 28.93%.

2,3,5,6,7,8-Hexahydro-4-cyano-3-oxoisoquinoline-1-thiol(2). A sample of compound(1) (2.2 gm, 0.0] mole) was refluxed with 50 ml 1N NaOH for 4 hours, then the reaction mixture was allowed to

cool and acidified with conc. HCl, the solid product was separated by filtration, washed well with water and recrystallized from benzene-ethanol mixture to give white crystals of compound(2) in 80% yield, m.p. $> 300^{\circ}$ C.

Anal. Calcd: $C_{10}H_{10}N_2OS$: C, 58.25; H, 4.85; N, 13.59; S, 15.53% Found: C, 58.38; H, 5.02; N, 13.38; S, 15.42%. IR, at 3120 cm⁻¹ (NH), 2500–2400 cm⁻¹ (SH), 2220 cm⁻¹ (C=N) and at 1700–1650 cm⁻¹ (C=O).

H¹ NMR in DMSO-d₆ at δ 1.3–1.8 (m, 4H, 2 β CH₂) and δ 2.3–2.6 (m, 4H, 2 α CH₂) of cyclohexane ring, δ 4.7 proad peak (1H, SH) and 7.4 (s, 1H, NH), MS: m/z = 206 (M⁺).

5,6,7,8-Tetrahydro-3-alkoxy-1-alkylthioisoquinoline-4-carbonitrile(3). General procedure: A mixture of compound(2) (2 gm, 0.01 mole) alkyliodide (0.02 mole) and sodium acetate (2.00 gm) in 30 ml ethanol was refluxed for 4 hours, then allowed to cool, and poured into water. The solid product was filtered off and recrystallized from ethanol to give compounds (3_a) and (3_b).

3a, $R = CH_3$, produced in 70% yield m.p. 93-94°C.

Anal. Calcd.: $C_{12}H_{14}N_2OS$: C, 61.53; H, 5.98; N, 11.96; S, 13.67%. Found: C, 61.64; H, 6.21; N, 11.87; S, 13.57%. H¹ NMR in CDCl₃ at δ 1.4–18 (m) (4H, 2 β CH₂), δ 2.2–2.4 (m, 2H, α CH₂) and 2.7–2.9 (m, 2H, α CH₂) of cyclohexane) at 2.5 (s, 3H, SCH₃) and at 3.9 (s, 3H, OCH₃). MS: m/z = 234(M⁺).

3b, $R = C_2H_6$ produced in 65% yield, m.p. 262°C.

Anal. Calcd.: $C_{14}H_{18}N_2OS$: C, 64.12; H, 6.8; N, 10.68; S, 12.21%. Found: C, 64.32; H, 7.02; N, 10.58; S, 12.00%.

2,3,7,8,9,10-Hexahydro-3,5-dioxo-5H-thiazolo[2,3-a]isoquinoline-6-carbonitrile(4). A mixture of compound(2) (2 gm, 0.01 mole), chloroacetic acid (1 gm, 0.01 mole) and fused sodium acetate (1.7 gm, 0.02 mole) in 30 ml ethanol was refluxed for 12 hours, then the reaction mixture was allowed to cool and poured into water. The solid product was filtered off and recrystallized from ethanol to give white crystals of (4) in 87% yield m.p. 198-200°C.

Anal. Calcd.: $C_{12}H_{10}N_2O_2S$: C, 58.78; H, 4.06; N, 11.38; S, 13.00%. Found: C, 58.78; H, 4.18; N, 11.22; S, 12.96%. IR, at 2220 cm⁻¹ (CN), 1720 cm⁻¹ (C=O) and at 1660 cm⁻¹ (C=O) H¹ NMR in DMSO- d_6 , at δ 1.5-1.9 (m, 4H, 2 β CH₂), δ 2.1-2.3 (m, 2H, α CH₂), and 2.6-2.8 (m, 2H, α CH₂) of cyclohexane and at 4.00 (s, 2H, CH₂) of thiazol ring. MS: m/z = 246(M⁺).

3,4,8,9,10,11-Hexahydro-4,6-dioxo-2H,6H[1,3]thiazino[2,3-a]-7-carbonitrile(5). A mixture of compound(2) (2 gm, 0.01 mole), β -bromopropionic acid (1.5 gm, 0.01 mole) and fused sodium acetate (1.7 gm, 0.02 mole) in 30 ml ethanol was refluxed for 12 hours, then the reaction mixture was allowed to cool and poured into water. The solid product was separated by filtration and recrystallized from ethanol to give white crystals of compound(5) in 70% yield, m.p. 226-28°C.

Anal. Calcd.: $C_{13}H_{12}N_2O_2S$: C, 60.00; H, 4.61; N, 10.72; S, 12.30%. Found: C, 60.28; H, 4.81; N, 10.59; S, 12.28%. IR, 2220 cm⁻¹ (C=N), 1720 cm⁻¹ (C=O), and 1670 cm⁻¹ (C=O). MS: $m/z = 260(M^+)$.

2,3,5,6,7,8-Hexahydro-1-aroylmethylthio-3-oxoisoquinoline-4-carbonitrile(6). General procedure: A mixture of compound(2) (2 gm, 0.01 mole) and phenacyl bromide or any of its derivatives in 30 ml ethanol was refluxed for 4 hours, then the reaction mixture was allowed to cool. The solid product was filtered off and recrystallized from ethanol to produce compounds (6a-c).

6a, Ar=C₆H₅ produced in 78% yield, m.p. 160°C.

Anal. Calcd.: $C_{18}H_{16}N_2O_2S$: 66.66; H, 4.93; N, 8.64; S, 9.87%. Found: C, 66.88; H, 5.08; N, 8.52; S, 9.96%. IR, at 3430 cm⁻¹ (NH), 2200 cm⁻¹ (C=N), and 1700–1650 cm⁻¹ (C=O) H¹ NMR in CDCl₃ at δ 1.5–1.8 (m, 4H, 2 β CH₂), δ 2.2–2.4 (m, 2H, α CH₂), δ 2.7–2.9 (m, 2H, α CH₂) of cyclohexane, δ 4.5 (s, 2H, S-CH₂), δ 7.5 (s, 1H, NH) and δ 6.9–7.4, (m, 5H, Ar–H). MS: m/z = 324(M⁺).

6b, Ar= C_6H_4Cl produced in 70% yield, m.p. 233-35°C.

Anal. Calcd.: $C_{18}H_{15}CINO_2S$: C, 60.25; H, 4.18; Cl, 9.90; N, 7.81; S, 8.92%. Found: C, 60.38; H, 4.33; Cl, 9.87; N, 7.62; S, 9.12%.

6c, Ar=C₆H₄Br, produced in 50% yield, m.p. 180°C).

Anal. Calcd.: C₁₈H₁₅BrN₂O₂S: C, 53.59; H, 3.72; Br, 19.85; N, 6.94; S, 7.94%.

Found: C, 53.72; H, 3.18; Br, 19.75. N, 6.87; S, 8.08%.

7,8,9,10-Tetrahydro-5-oxo-3-phenyl-5H-thiazolo[2,3-a]isoquinoline-6-carbonitrile(7). A sample of 6a was heated over its melting point for 1 hour, then allowed to cool, ethanol (30 ml) was added and refluxed for 2 hours. The solid product was separated by filtration and recrystallized from acetic acid to give pale yellow crystals of compound(7) in 70% yield, m.p. 185°C.

Anal. Calcd.: $C_{18}H_{14}N_2OS$: C, 70.58; H, 4.57; N, 9.15; S, 10.45%. Found: C, 70.34; H, 4.52; N, 9.22; S, 10.18%. IR, 2200 cm⁻¹ (C=N), 1670 cm⁻¹ (C=O).

5,6,7,8-Tetrahydro-3-aroylmethoxy-1-aroylmethylthioisoquinoline-4-carbonitrile(8). General procedure: A mixture of compound(2) (2 gm, 0.01 mole), phenacyl bromide or any of its derivatives (0.02 mole) and fused sodium acetate (0.02 mole) was refluxed in 30 ml ethanol for 6 hours, the solid product was filtered off washed well with water and recrystallized from ethanol to produce compounds 8a-c.

8a Ar=C₆H₅ produced in 70% yield, m.p. 213-15°C.

Anal. Calcd.: $C_{26}H_{22}N_2O_3S$: C, 70.58; H, 4.97; N, 6.33; S, 7.23%. Found: C, 70.82; H, 4.89; N, 6.12; S, 7.54%. IR at 220 cm⁻¹ (C=N), 1720 cm⁻¹ and at 1690 cm⁻¹ (2CO).

H¹ NMR in CDCl₃ at δ 1.5–1.8 (m, 4H, 2 β CH₂), δ 2.2–2.4 (m, 2H, α CH₂) and δ 2.6–2.8(m, 2H, α CH₂) of cyclohexane ring, δ 4.3 (s, 2H, SCH₂), 5.4 (s, 2H, OCH₂) and δ 7–7.5 (m) (10H, ArH). MS: m/z = 442(M⁺).

8b, Ar= C_6H_4Cl produced in 70% yield, m.p. 185-87°C.

Anal. Calcd.: $C_{26}H_{20}Cl_2N_2O_3S$: C, 61.17; H, 3.92; Cl, 13.72; N, 5.49; S, 6.27%. Found: C, 61.33; H, 4.08, Cl, 13.58, N, 5.62, S, 6.34%.

8c, Ar= C_6H_4Br , produced in 65% yield, m.p. > 300°C.

Anal. Calcd.: $C_{26}H_{20}Br_2N_2O_3S$: C, 52.00; H, 3.33; Br, 26.66; N, 4.66; S, 5.33%. Found: C, 52.28; H, 3.52; Br, 26.48; N, 4.48; S, 5.22%.

4,5,6,7-Tetrahydro-8-amino-3-benzoylmethylthio-9-benzoylfuro[2,3-b]isoquinoline(9). A mixture of 8a (0.9 gm, 0.002 mole) and potassium carbonate (2 gm) in 30 ml ethanol was refluxed for 8 hours then allowed to cool and poured into water. The solid product was filtered off and recrystallized from ethanol as yellow crystals in 70% yield m.p. 248-50°C.

Anal. Calcd.: $C_{26}H_{22}N_2O_3S$: C, 70.58; H, 4.97; N, 6.33; S, 7.23%. Found: C, 70.83; H, 5.18; N, 6.12; S, 7.08%.

IR, 3300, 3200 cm⁻¹ (NH₂), 1680 cm^{-1} (CO).

REFERENCES

- W. E. Bondinell, F. W. Chapin, G. R. Girard, C. Kaiser, A. J. Krog, A. M. Pavloff, M. S. Schwartz, J. S. Silvestri and P. D. Vaidya, J. Med. Chem. 23(5), 506-11 (1980).
- 2. W. E. Bodinell, R. G. Pendleton, U.S. Pat. 4,228,170, C.A. 94, 47155k (1981).
- 3. W. E. Bondinell and G. R. Girard, U.S. Pat 54,343 C.A. 95, 25072z (1981).
- 4. W. E. Bondinell and G. R. Girard, U.S. Pat 4,258,049 (C.A. 95, 43131k) (1981).
- 5. C. Kaiser and R. C. Pendleton, Intra-Science Chem. Rept., Vol. 8, No. 4 (1974).
- 6. K. Gewald, J. Prakt. Chem. 31, 205 (1966).